

Reg No.: \_\_\_\_\_

Name: \_\_\_\_\_

**APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY**  
FIRST SEMESTER B.TECH DEGREE EXAMINATION(2019 SCHEME), DECEMBER 2019

**Course Code: CYT100**

**Course Name: ENGINEERING CHEMISTRY**  
**(2019-Scheme)**



Max. Marks: 100

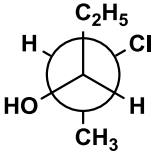
Duration: 3 Hours

**PART A**

*Answer all questions, each carries 3 marks.*

- 1 Calculate the equilibrium constant for the following reaction at 25°C:-  

$$\text{Fe}_{(s)} + \text{Cu}^{2+}_{(aq)} = \text{Fe}^{2+}_{(aq)} + \text{Cu}_{(s)}$$
 Given  $E^0_{\text{Fe}^{2+}/\text{Fe}} = -0,44 \text{ V}$ ,  $E^0_{\text{Cu}^{2+}/\text{Cu}} = 0.34 \text{ V}$  (3)
- 2 Give the electrochemical reaction taking place when an iron nail is dipped in dil.HCl.  $E^0_{\text{Fe}^{2+}/\text{Fe}} = -0,44 \text{ V}$ ,  $E^0_{\text{Fe}^{3+}/\text{Fe}} = -0,04 \text{ V}$ ,  $E^0_{\text{H}^+/\text{H}_2} = 0 \text{ V}$ . (3)
- 3 State and explain the law governing absorption of electromagnetic radiation by matter. Give any one limitation of this law. (3)
- 4 Which molecule will absorb at longest wavelength in UV? Explain. (3)
  - a) 
  - b) 
- 5 What are the classifications of chromatography based on physical state of mobile and stationary phases? (3)
- 6 Explain the synthesis of nanoparticles by chemical reduction. (3)
- 7 Write the IUPAC name and assign R/S notation. (3)
 


- 8 Write the different types of copolymers formed by the monomers A and B. (3)
- 9 Calculate the hardness of (i) 0.05 M  $\text{AlCl}_3$  and (ii) 0.04 N  $\text{MgCl}_2$ . (3)
- 10 What is the significance of measuring BOD of waste water? (3)

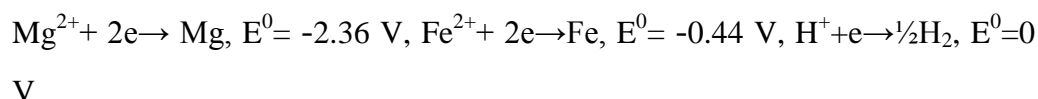
**PART B**

*Answer one full question from each module, each question carries 14 marks*

**Module-I**

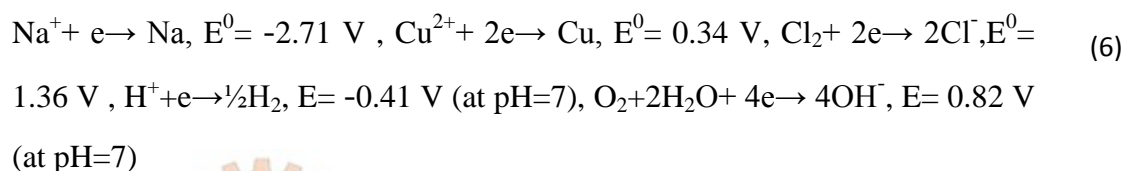
- 11 a) Explain the construction and working of a calomel electrode as a reference electrode. What is the variation in the potential of a calomel electrode with change in chloride ion concentration? (8)

- b) Why Mg corrodes in both acidic and alkaline oxygen deficient conditions, whereas Fe does not corrode in alkaline oxygen deficient condition? (6)



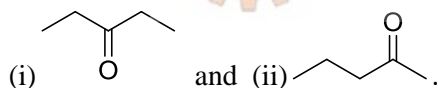
- 12 a) Write the construction, working and advantages of Li-ion cell. (8)

- b) What are the products of electrolysis at cathode and anode when NaCl solution is electrolysed using Cu electrodes.



### Module-II

- 13 a) Predict the number of signals, their relative positions and splitting pattern in the nmr spectrum of the following. (8)



- b) Compare the strengths of C-H bond and C=O bond if the absorption frequencies are  $3000\text{cm}^{-1}$  and  $1700\text{cm}^{-1}$  respectively. (6)
- 14 a) Give the instrumentation of UV spectrophotometer and explain the components in it. Comment on the role of conjugation in the wavelength of absorption with the help of examples. (8)
- b) Briefly explain the principle involved in MRI. Mention any two applications. (6)

### Module-III

- 15 a) Discuss in detail the Instrumentation of TG and DTA with neat sketch. (8)
- b) Discuss the various detectors used in GC and HPLC. (6)
- 16 a) Briefly explain the principle, instrumentation and applications of SEM. (8)
- b) Differentiate between TGA and DTA. (6)

### Module-IV

- 17 a) Draw and explain the conformational isomerism in ethane and butane. Draw the energy profile diagram. Which conformer is more stable in each case? (10)
- b) Explain the classification of conducting polymers. (4)
- 18 a) What is meant by conformational isomerism? Draw the *cis* and *trans* isomers of 1,4-dimethyl cyclohexane. In each case, mention the more stable conformer. (8)
- b) Brief out the basic principle, construction and working of OLED. (6)

### Module-V

- 19 a) Describe the various steps involved in sewage treatment. (10)
- b) Write any four disadvantages of hard water. (4)
- 20 a) Write the principle and procedure of estimation of permanent and temporary hardness of water by complexometric titration. (8)
- b) 50 mL sewage water sample after reaction with 20 mL of  $K_2Cr_2O_7$  required 12.4 mL of 0.2 N ferrous ammonium sulphate solution. For blank titration 20 mL  $K_2Cr_2O_7$  required 20.4 mL of 0.2 N ferrous ammonium sulphate solution. (6)
- Calculate the COD of the sample.

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